Attachment E Plan for Evaluating SWIFT Soil Aquifer Treatment

# Plan for Evaluating SWIFT Soil Aquifer Treatment

Prepared for

United States Environmental Protection Agency

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Prepared by



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# Acronyms and Abbreviations

contaminant of emerging concern

AWTP advanced water treatment plant

BDE brominated diphenyl ether

CCL candidate contaminant list

DBP disinfection by-product

DO dissolved oxygen

CEC

DOC dissolved organic carbon

EPA U.S. Environmental Protection Agency

fbg feet below grade

HAA5 haloacetic acids (monochloroacetic, dichloroacetic, trichloroacetic, monobromoacetic,

and dibromoacetic)

HAA haloacetic acid

HRSD Hampton Roads Sanitation District

LPA Lower Potomac aquifer

m<sup>3</sup> cubic meter(s)

MAR managed aquifer recharge

mgd million gallon(s) per day

mg/L milligram(s) per liter

mL milliliter(s)

mL/min milliliter(s) per minute

MPA Middle Potomac aquifer

MPN most probable number

NDMA N-Nitrosodimethylamine

ORP oxidation-reduction potential

PAS Potomac Aquifer System

pfu plaque forming unit

PMMoV pepper mild mottle virus

PVC polyvinyl chloride

PVFD polyvinylidene fluoride

RNA ribonucleic acid

SAT soil aguifer treatment

SWIFT Sustainable Water Initiative for Tomorrow

SWIFTRC SWIFT Research Center

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TDS total dissolved solids

THM trihalomethane

TKN total kjeldahl nitrogen
TOC total organic carbon

UCMR Unregulated Contaminant Monitoring Rule

UIC Underground Injection Control

UPA Upper Potomac aquifer
UVD Ultraviolet disinfection

WWTP wastewater treatment plant

# Introduction

# 1.1 Background

The Hampton Roads Sanitation District (HRSD) Sustainable Water Initiative for Tomorrow (SWIFT) will add multiple advanced water treatment processes to select HRSD wastewater treatment facilities to produce a highly treated water (SWIFT Water) that exceeds drinking water standards and is compatible with the receiving aquifer. Secondary effluent from up to seven of HRSD's existing treatment facilities will be treated at SWIFT facilities and SWIFT Water will be recharged into the Potomac Aquifer System (PAS) to counter depleting aquifer levels. At full-scale, HRSD intends to recharge over 100 million gallons per day (mgd) of SWIFT Water that will significantly reduce the nutrient load to the sensitive Chesapeake Bay and provide significant benefit to the region by limiting saltwater intrusion, reducing land subsidence, and providing a sustainable source of groundwater, a necessity for continued economic expansion in the region.

The SWIFT Research Center (SWIFTRC) involves a nominal 1 mgd advanced treatment facility and injection well located at the Nansemond Treatment Plant (Suffolk, Virginia) that will begin production and recharge in spring 2018. The primary purpose of the SWIFTRC is to demonstrate at a meaningful scale that advanced treatment will produce SWIFT Water that meets primary drinking water standards and is compatible with the groundwater chemistry and minerals composing the PAS. HRSD will collect at least 18 months of operational data to inform and optimize the design and construction and to define permitting requirements for the full-scale SWIFT facilities.

### 1.2 Purpose

This document is Attachment E within the Underground Injection Control (UIC) Inventory Information Package ("UIC Inventory"). The purpose of this document is to characterize soil aquifer treatment (SAT) through column testing experiments simulating managed aquifer recharge (MAR) operations and describes field scale studies during the operations.

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# Bibliography of Previous SAT Studies

# 2.1 SAT Pathogen Removal

A key consideration for the implementation of indirect potable water reuse is the removal and/or inactivation of pathogens through the advanced treatment process and subsequent aquifer recharge. The SWIFT RC Advanced Water Treatment Plant (AWTP) at Nansemond has been designed to provide log reduction values of viruses, *Cryptosporidium*, and *Giardia* to exceed various regulatory requirements. The purpose of this technical memorandum is to demonstrate that additional pathogen removal credits should be considered for SAT as part of the overall public health protection barriers included in the SWIFTRC AWTP treatment process.

#### 2.1.1 Regulations and Research Supporting SAT Pathogen Removal

California, an early developer of regulations on recycled water, requires 12-log reduction of viruses and 10-log reduction of both *Cryptosporidium* and *Giardia* for potable reuse applications. California regulations published in July of 2015 state that recycled municipal wastewater is credited with 1-log virus removal for each month that the recycled water is in the aquifer up to a 6-log reduction. Prior to aquifer recharge, the recycled water must also meet the definition of filtered wastewater and disinfected tertiary recycled water outlined in the California Regulations. The SWIFT Water at HRSD will meet these criteria. To receive credit for virus reduction in the aquifer, residence time in the aquifer must be validated with a tracer study, starting within three months of AWTP operation. The project sponsor must also:

- Provide documentation of an alternative water source for drinking water well users in case of treatment failure
- Collect at least four water samples, at least one per quarter, from affected aquifers prior to AWTP operation
- Provide a map of the project site including all drinking water and monitoring wells within a two year travel time and including potential future wells
- Provide an engineering report with a hydrogeological assessment of the AWTP setting including
  properties and extent of affected aquifers, quarterly groundwater elevation contours and calculated
  flow directions and gradients. Maps and assessments must be based on quarterly evaluations to
  capture seasonal changes
- Demonstrate treatment processes are operating as intended
- Construct at least two monitoring wells
- Regularly analyze AWTP effluent and monitoring well samples for pathogens, total organic carbon (TOC), nitrogen compounds, contaminants, and chemicals with primary and secondary monochloramines
- Must report failure to meet required pathogen reduction

### 2.1.2 SAT Pathogen Research

Significant effort has been devoted to studying the effectiveness of SAT for pathogen removal. Studies range from pathogen removal in laboratory columns to full scale field studies at MAR sites. Pathogen removal depends on many factors including travel time in the aquifer, distance travelled in the aquifer,

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velocity, aquifer soil texture, microbe type, and water chemistry. Table 2.1 shows that significant virus removal has been demonstrated using SAT in a variety of aquifer and laboratory conditions.

Table 2-1. Summary of SAT Pathogen Studies

| Reference                  | Study  | Soil/Aquifer Type   | Microbes  | Virus Removal   |
|----------------------------|--|---|---|---|
| Elkayam et<br>al., 2016    | 30 years of aquifer recharge                                       |   | Fecal coliform,<br>enteroviruses                      | Fecal coliform: >5-log removal after 17 days, complete removal at reclamation wells (960 day travel time) for 829/831 tests from 1980 to 2012, no detection since 1995  |
| ·                          | in Israel  |   |   | Enteroviruses: complete removal in 54/57 tests (~370 day travel time), no detection since 2001  |
|                            |  |   | E. coli, Bacteroides, coliphage, PMMoV,               | Log reduction over 65 meters of riverbank filtration:   |
| Verbyla et<br>al., 2016    | Two riverbank<br>filtration sites in<br>Bolivia                    | Riverbank   | rotavirus,<br>adenovirus, Giardia,<br>Cryptosporidium | E. coli: 3.8, Bacteroides: 5.5, coliphage: 2.0, PMMoV: 2.9, rotavirus: >3.5, adenovirus: >3.5, Giardia: 2.1, Cryptosporidium: 1.7   |
| Sidhu et al.,<br>2015      | Diffusion<br>chambers in four<br>Australian<br>aquifer systems     | Confined limestone aquifer, unconfined superficial aquifer of iron-coated siliceous sand- carbonated cemented sand, and unconfined quaternary sand and gravel aquifer | Bacterial<br>pathogens, oocysts,<br>enteric viruses   | Time for 1-log removal: Bacteria: < 3 days Oocysts: < 120 days Enteric viruses: 18 to > 200 days  |
| Betancourt<br>et al., 2014 | Three aquifer recharge systems: Arizona, California, and Colorado  | Colorado<br>Riverbank site:<br>alluvial sand with<br>some gravel and<br>silts, SAT sites: and<br>coarse sand/sandy<br>gravel  | Adenoviruses,<br>enteroviruses, Aichi<br>virus, PMMoV | Viruses removed below detection limit—log removal quantification difficult: Arizona site: > 3.42-log removal of all viruses in ~14 days California site: > 1.05-log removal of adenovirus and non-detect of other viruses in 0.5-128.5 days Colorado site: >0.7 log removal of adenovirus, > 1.15-log removal of enterovirus, > 2.49-log removal of Aichi virus and > 1.92-log removal of PMMoV in 5 days |
| Santamaria<br>et al., 2013 | In-situ 4 meter x<br>2.5 meter<br>lysimeter and 2-<br>meter column | Sandy soil  | Cryptosporidium                                       | 5-log removal in 2.4 days in lysimeter; 4-log removal in 0.8 days in 2-meter column   |
| Page et al.,<br>2010       | Diffusion<br>chamber in<br>Australian<br>aquifer                   |   | Rotavirus,<br>Cryptosporidium,<br>Campylobacter       | Rotavirus: 0.0055-log/day, <i>Cryptosporidium</i> : 0.012 log/day, <i>Campylobacter</i> : total 5.6-log removal   |
| Fox et al.,<br>2006        | Tracer study in<br>California aquifer                              | Shallow vadose zone aquifer   | Bacteriophages  | 7-log removal in 100 feet of subsurface travel  |
| Hijnen et al.<br>2005      | Laboratory 0.5<br>meter columns                                    | Sandy soil and gravel soil  | MS2, E. coli, C. perfringens, C.                      | Sandy aquifer soil (column at 0.5 meter per day):   |
|                            | with soil from an  |   | parvum, G.  | MS2: 3.3, E. coli: 4.7, C. perfringens: >5, C.  |

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Table 2-1. Summary of SAT Pathogen Studies

| Reference                | Study  | Soil/Aquifer Type        | Microbes  | Virus Removal   |
|--------------------------|--|--------------------------|---|---|
|                          | aquifer recharge   |                          | intestinalis                                      | parvum: 3.9, G. intestinalis: 6.2   |
|                          | site and a<br>riverbank<br>filtration site in                |                          |   | Gravel river aquifer soil (column at 0.9 meter per day):  |
|                          | the Netherlands  |                          |   | MS2: 3.4, E. coli: 4.8, C. perfringens: >2.4, C. parvum: >6.7, G. intestinalis: >7.4  |
| Anders et al., 2004      | Tracer study in<br>California aquifer                        | Fine-coarse sand aquifer | MS2 and PRD1                                      | 1998 experiment: 0.37-log units/meter MS2;<br>0.55-log units/meter PRD1. 2000 experiment:<br>0.83-log units/meter MS2; 3.0-log units/meter<br>PRD1    |
| Quanrud et al., 2002     | WWTP<br>secondary<br>effluent in 1-<br>meter soil<br>columns | River sand or sandy loam | Coliphages and poliovirus                         | Coliphage: expected 2-log removal in 17.5 hours, 3-log removal in 26 hours  |
| Tufenkji et<br>al., 2002 | Several riverbank<br>filtration sites in<br>the Netherlands  | Riverbank                | Bacteriophages,<br>coliphages, enteric<br>viruses | Average log removal in 15-63 days:<br>RNA bacteriophages: 6.0<br>Enteric viruses: 4.0<br>Coliforms: 5.0<br>Clostridia: 3.3<br>Fecal streptococci: 3.3 |

Note:

m = meter(s)

PMMoV = pepper mild mottle virus

RNA = ribonucleic acid

WWTP = wastewater treatment plant

A paper published in 2009 by Liping Pang compiles results from over 150 field and laboratory experiments. Table 2.2 shows average log removal rates per meter for *E. coli*, enterococci, fecal coliforms, fecal streptococci, and *Salmonella* phage bacteria from lysometer studies. Averages include experiments conducted in various locations with various microbial sources and soil types.

# 2.2 SAT Organics Removal

Soil column experiments have been used to study the fate and transport of different compounds present in water to be used for aquifer replenishment. The focus of most studies has been on contaminants of emerging concern (CECs). Banzhaf et al. (2012)

Table 2-2. Summary of Lysimeter Studies compiled by Pang, 2009

| Microbe            | Average Log<br>Removal/Meter |
|--------------------|------------------------------|
| E. coli            | 0.59                         |
| Enterococci        | 0.53                         |
| Fecal coliforms    | 3.25                         |
| Fecal streptococci | 4.02                         |
| Salmonella phage   | 2.34                         |

studied sorption and biodegradation of carbamazepine, sulfamethoxazole and diclofenac. Strauss et al. (2011) and Fan et al. (2011) observed removal mechanisms for sulfamethoxazole and its metabolites. Other CECs that have been considered in soil column experiments are Bisphenol A, 17  $\beta$ -estradiol, 17  $\alpha$  – ethynyl estradiol (Patterson et al., 2010), primidone, atenolol, meprobamate (Burke et al., 2013), perfluoroalkyl acid (McKenzie et al., 2016) along with dissolved organic carbon (DOC) and organic halide (Quanrud et al., 1996). Few have also looked at the effects of using different electron acceptors (Nay et al., 1999) and substrates (Hebig et al., 2017) on the transport and degradation of these CECs. A summary of relevant papers is available in Table 2.3 below.

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| Title   | Citation  | Diam (m)    | Height (m)     | Flow Rate/  | Flow Rate (mL/min)  | Effective            |                            | Travel Time(day) | Sampling frequency   | Contaminants  | Mechanism tested  | Comments   |
|---|---|-------------|----------------|---|---|----------------------|----------------------------|------------------|--|---|---|--|
|   | Citation  | Diam (m)    | rieigiit (iii) | velocity (reported)   | now nate (mr./mm)   | Porosity             | velocity (cili/u)          | Traver rime(uay) | Sampling frequency   | Containmants  | Wechanism tested  | Comments   |
| Sorption behavior of<br>20 wastewater<br>originated micropollutants<br>in<br>groundwater -<br>Column experiments<br>with pharmaceutical<br>residues and<br>industrial agents              | Burke, et al. (2013).<br>Journal of contaminant<br>hydrology , 154 , 29-41.                 | 0.1         | 0.3            | 5.6E-9 m3/s   | 0.336   | 0.45                 | 13.7                       | 2.19             | Conservative tracer:<br>4 hours<br>Contaminants: 30 hours  | Diazepam, Oxazepam,<br>Primidone, PEMA, Atenolol,<br>FAA, AAA, AMPH,<br>Propanolol, Sotalol,<br>Metoprolol, Totalol,<br>Tolyltriazole, Phenacetine,<br>Methyl-phenacetine                                       | Sorption /<br>Desorption; biological degradation<br>neglected | Column was conditioned with tap water until constant conditions of phoxygen and temperature reached. Influent with contaminants and conservative tracer (NaCl) was then passed through column for 5 days Inflow was switched to tap water and samples taken for 7 days.  |
| Redox sensitivity and<br>mobility of pharmaceutica<br>compounds in a low flow<br>column experiment  |   | 0.136       | 0.351          | 14.1 mL/h   | 0.235   | 0.41                 | 5.7                        | 6.18             | Contaminants:<br>collected every 3 hours,<br>every 5th sample was<br>analyzed  | Carbamazepine,<br>Sulfamethoxazole,<br>diclofenac   | Sorption and degradation                                      | Conditioning period with nonspiked water - 2.5 months. Concentration tested in the range of 175 - 852 ng/L. Influent was spiked with the contaminants. The aim was to test the Influence of nitrate concentration on breakthrough behavior. CXFIT code was used for hydraulic modeling Result: Carbamazepine seemed degradable when fraction of organic carbon (foc) of sediment was high. Silfamethoxazole is sensitive to nitrate reducing redox conditions. |
| Guidelining protocol for so<br>column experiments<br>assessing fate and<br>transport of trace organic   | Hernández<br>Amphos 21: Ester   | 0.1-0.2     | 1.5 - 2.5      | not reported  | not reported  | Not reported         | not reported               | not reported     | not reported   | not reported  | not reported  | Guideline/Review   |
| Fate of organics during column studies of soil aquifer treatment  | Quanrud, et al. (1996).<br>Journal of<br>Environmental<br>Engineering, 122(4), 314-<br>321. | 0.0826      | 1.00           | 200 ml/h<br>in wet cycle  | 3.33<br>in wet cycle  | 0.46                 | 194.5                      | 0.51             | not reported   | DOC, adsorbable organic halide  | sorption, microbial degradation                               | Columns were subjected to alternating wet and dry cycles each of 7 days Unsaturated conditions. Result: DOC was removed by biodegradation, AOX was removed by sorption.  |
| Use of column<br>experiments to investigate<br>the fate of organic<br>micropollutants - review  | System Sciences, 20(9),   | 0.02 - 0.36 | 0.05 - 2.4     | not reported  | not reported  | not reported         | 4 - 348 cm/d<br>(velocity) | 0.7 - 1.25       | not reported   | not reported  | not reported  | Review Paper   |
| Do Pharmaceuticals,<br>Pathogens and other<br>organic waste water<br>componds persist when<br>wastewater is used for<br>recharge?   | Cordy, et al.<br>(2004) Groundwater<br>Monitoring &<br>Remediation , 24(2), 58<br>69.       | 0.325       | 2.1            | 5.3 cm/d (q=Q/A)  | 3.1   | 0.38                 | 13.9                       | 15.06            | not reported   | vet and human antibiotics<br>prescription drugs,<br>nonprescription drugs,<br>household and industrial<br>chemicals, steroids and<br>reproductive hormones  | not reported  | Experiment designed to approximate recharge conditions similar to those of a wetting cycle in recharge spreading basin. 200 L of secondary effluent discharged on the surface and allowed to infiltrate. The effluent from the SAT columns are tested for contaminants to check their persistence.   |
| Sorption and<br>biodegradation of organic<br>micropollutants during rive<br>bank filtration; A laborator<br>column study  | er (2014). Water  | 0.36        | 1.00           | v=2.4-3.2 m/d   | not reported  | 0.31-0.42            | 240-320                    | 0.31-0.42        | influent and effluent<br>measured at 6 different<br>time periods in a month<br>(10 h, 34 h, 58.5 h, 80 h, 1<br>wk, 4 hrs)  | ibuprofen, ketoprofen,<br>gemfibrozil, acetaminophen,<br>trimethoprim, caffeine,<br>propranolol, metoprolol,<br>atrazin, carbamazepine,<br>phenytoin,<br>sulfamethoxazole,<br>hydrochlorothiażide,<br>linomycin | sorption and bio-<br>degradation                              | Aim of the study was to obseve if physico-chemical parameters such as hydrophobicity, charge and molecular weight affected biodegradation rates. Columns were fed with surface water from local canal; Adaptation period of 4 months.  |
| Effects of pH and<br>manure on transport of<br>sulfonamide antibiotics in<br>soil   | Strauss, et al. (2011).<br>Journal of<br>environmental<br>quality, 40(5), 1652-<br>1660.    | 0.052       | 0.3            | 0.2 ml/min - before<br>reaching saturation point;<br>1.44 ml/min after<br>saturation ; Darcy velocity -<br>4 cm/h | 0.2 - before<br>reaching<br>saturation point;<br>1.44 after<br>saturation                   | 0.38-0.40            | 248.0                      | 0.12             | not reported   | sulfamethoxazole,<br>sulfamethazine,<br>sulfadimethoxine  | not reported  | Aim of the study was to test the hypothesis that manure as cosolute an pH influence sulfonamide transport. Breakthrough curves of tracer and sulfonamides at different pHs was modeled using Hydrus 1-D.   |
| Sorption, fate and mobility of sulfonamides in soils  |   | 0.084       | 0.15           | 19.8-39.5 cm/h  | constant head<br>tank used for<br>constant<br>downward flow.<br>Flow rate not<br>mentioned. | Not reported         | 480-960                    | 0.029-0.036      | column effluent collected<br>every 2 mins.   | sulfonamides  | sorption  | In order to describe the fate and transport of Sulfamethazine (SMZ) and its metabolite under steady-state flow in a homogeneous soil, a two-sit chemical nonequilibrium transport model was used. Duration of column experiment: 6 h. soil extracts from column analyzed to analyze sorption affinity. Result: SMZ had low sorption affinity and all sorption was reversible.  |
| Transport of primidone,<br>carbamazepine, and<br>sulfamethoxazone in<br>hermally treated sediment<br>lab column experiments   |   | 0.135       | 0.35           | 1.0-1.3 ml/min for<br>four different column<br>experiments  | 1.0 - 1.3 for<br>four different<br>column<br>experiments                                    | 32% of vol<br>(0.32) | 28.8-37.4                  | 0.93-1.22        | total of 20 samples of<br>effluent collected per test.<br>sample collected every<br>hour. Sample volume - 75<br>ml   | carbamazepine,<br>sulfamethoxazole,<br>primidone  | degradation, sorption   | Column study: 4 column expriments (untreated sediment, pretreated sediment at different temperatures) Results: all three compounds showed similar transport behavior of conservative tracer. Carbamazepin and Primidone are retarded in the presence of organic matter. Order of decreasing retardation CB2>PMD>SMX  |
| Fate of nine recycled wate<br>trace organic contaminant<br>and metal(oids) during<br>managed aquifer recharge<br>into a anaerobic aquifer:<br>column studies                              | s Patterson, et al.<br>(2010).Water<br>research, 44(5), 1471-                               | 0.145       | 2.00           | 360 ml/day;<br>velocity - 5.2 cm/day  | 0.25  | 0.42                 | 5.2                        | 38.46            | not reported   | Bisphenol A, 17 b estradiol,<br>17 a -ethynylestradiol,<br>carbamazepine, N-<br>nitrosomorpholine, iohexol  | retardation, degradation                                      | Setup: 17 sampling ports along the column; Influent is RO treated water<br>Results: Anaerobic consitions were good for the degradation of<br>bisphenol, estradiol, ethynylestradiol. Carbamazepine, oxazepam did no<br>degrade readily.  |
| Transport of<br>Pharmaceutically Active<br>Compounds in Saturated<br>Lab Columns  |   | 0.14        | 0.35           | 5.9 x 10^-5 m3/hour<br>velocity - 0.3 m/d; specific<br>discharge - 0.097 m/day                                    | 0.98  | 0.32                 | 30.0                       | 1.17             | Sample volume - 25 ml; pH,<br>temp, O2 and saturation<br>measured every 10 mins,   | clofibric acid,<br>propyphenazone, diclofenac   | degradation, sorption   | The aim of the study was to study transport behavior of contaminants. The equilibration time S days; influent with tracer and PPCPs was passed through the column for 10 days; Column flushed for days   |
| Transport behavior of the pharmaceutical compound carbamazepine, sulfamethoxazole, gemfibrozil, ibuprofen, and naproxen, and the lifestyle drug caffeine, in saturated laboratory columns | Hebig, et al. (2017).<br>Science of The Total<br>Environment, 590, 708-<br>719.             | 0.076       | 0.41           | not reported  | 0.095   | 0.27-0.33            | 10.0                       | 4.10             | 68 samples collected at<br>intervals varying between 5<br>and 230 h  | Naproxen, Gemfibrozil,<br>Ibuprofen, caffeine,<br>sulfamethoxazole,<br>carbamazepine  | retardation, degradation                                      | Objective: Study the effect of substrates on the transport of micropollutants. Three columns used: iron clad sand, organic carbon sand, long point sediment; Breakthrough curves of compounds plotted. CXTFITT model used to determine Retardation factor, Dispersivity. Result: Sulfamethoxa  |
| Effects of Chemical<br>Oxidants on Perfluoroalky<br>Acid Transport in<br>One-Dimensional Porous<br>Media Columns  |   | 0.025       | 0.08           | v=83.4 cm/d   | 0.085   | 0.48                 | 83.4                       | 0.10             | PFAA: 18-27 samples collected per column per phase; plt samples: 6-7 samples per column per phase; TOC: 3 samples per column per phase; Metals: 3 samples per column per phase; Metals: 3 samples per column per phase | perfluoroalkyl acid (PFAs)  | sorption, degradation,  | Columns were loaded with (11) perfluoroalkyl acids (PFAAs) amounting t ~30 pore volumes. Chemical oxidants (permanganate and activated persulfate) were added later to evaluate mass/concentration reduction   |
| Fate and behavior of organic compounds in an artificial saturated subsoil under controlled redox conditions: The sequentia soil column system   | Biodegradation, 10, 75-   | 0.025       | 0.16           | Q=60mL/d  | 0.0417  | Not reported         | 40.7                       | 0.39             | recovery time of about 12<br>h between single samples<br>and a minimum recovery<br>time of 5 days undisturbed<br>operation were allowed<br>between sample series   | perchloroethene, 1,1-<br>dichloroethene, 1,4-<br>dichlorobenzene, 2,4-<br>dichlorophenol, 2-<br>nitrophenol, benzene,<br>toluene, naphthalene   | redox, biodegradation,  | Sequential Columns (4) - Travel Time through system ~ 1.5 d; Electron acceptors varied in each column (CO2, SO4, NO3, and O2). Chlorinatec VOCs and non-chlor VOCs added in solution. Column experiments were conducted in phases (i) first sorption (ii) pulse oxidation (iii) extended oxidation (iv) first desorption (v) second sorption (vi) second desorption  |

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### 2.3 Use of Soil Column Studies for SAT Investigations

Column set up, dimensions and operational procedures vary widely among different experimental studies. The diameters of the columns reported in literature range from 0.02 - 0.36 meters and the lengths are in the range of 0.05 – 2.4 meters (Banzhaf et al., 2016). In order to prevent preferential flow paths or sidewall flows, a diameter to length ratio of 1:4 is recommended (Gibert et al., 2015). The flow rate through the columns has varied from 0.04 milliliter per minute (mL/min) (Nay et al., 1999) to 3.33 mL/min (Quanrud et al., 1996), flow rates were chosen based on the desired travel times to be simulated. The column casings are commonly made up of materials such as stainless steel, glass and transparent polyvinyl chloride (PVC). Peristaltic pumps have been typically used to mechanize a pressurized upflow to let the influent move up the column. Upflow is preferred, particularly for saturated soil column tests, as it allows air bubbles to escape from the top. Packed soil columns are preferred over monoliths, which are undisturbed soil columns extracted from site, in column studies. Packed soil columns ensure homogeneity and can be constructed with bulk density similar to the natural site conditions (Lewis et al., 2010). After the construction of the soil columns and before beginning the experiment, columns are allowed to stabilize which is done by flushing with uncontaminated and/or non-spiked water. This adaptation period has been reported to be anywhere from 5 days (Scheytt et al., 2004) to 4 months (Bertelkamp et al., 2014). A conservative tracer is passed through the column. Tracer is a non-reactive compound that is not subject to sorption or biotic/abiotic transformation and is used to determine the boundary and initial conditions of the soil column (Lewis et al., 2010). Bromide is the most commonly used conservative tracer. Sodium Chloride and Lithium Chloride are other compounds that have been used for tracer tests (Burke et al., 2013; Schyett et al., 2004).

Feed water with the contaminants of interest is passed through the column and samples of the column effluent collected for analysis. Pumped feed water flow rate is used to control hydraulic retention time within the SAT column. The effluent concentrations are typically plotted against time or pore volume to produce a breakthrough curve and compared to the breakthrough curve of conservative tracer to detect any differences in their transport behavior. Numerical modeling has also been used in various studies to estimate parameters such as retardation factor, rate of biodegradation, and dispersivity of compounds. PHREEQC, CXTFIT and HYDRUS-1D are some of the software that are available and are applicable for modeling of one dimensional flow characteristics (Gibert et al., 2015). Muller et al. (2013) used CXTFIT to produce breakthrough curves of carabamazepine, primidone and sulfamethoxazole and determine the transport behavior of the compounds. The retardation of Carbamazepine and primidone were observed to increase in the presence of organic matter while sulfamethoxazole was the least removed compound among the three. On the other hand, Hebig et al.'s (2017) study shows that sulfamethoxazole gets removed in redox reactions. Hebig et al. used CXTFIT model to estimate retardation factors and dispersivity of sulfamethoxazole, ibuprofen, and other CECs. Soil column studies, therefore, provide meaningful information regarding the transport and degradation behaviors of compounds and the conditions under which their removal can be optimized. These considerations are significant for the full-scale implementation of managed aquifer replenishment.

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# Soil Column Testing

# 3.1 Project Introduction and Objectives

Soil column testing was used to evaluate and quantify the benefit of SAT in terms of pathogens, organics and CECs, disinfection by-products (DPBs), and nitrogen species. Phase I involved constructing four soil columns within the SWIFT Pilot facility at the HRSD York River Treatment Plant. The soil columns were fed pilot effluent (Ultraviolet disinfection effluent) from the SWIFT advanced treatment pilot facility. Subject to the results from Phase I, Phase II work will involve continued soil column work located at the SWIFTRC and fed SWIFT Water from the demonstration process.

The specific objectives of the soil column work included the following:

- Evaluate the removal of pathogens and pathogen indicators by SAT, with specific focus on confirming at least 1 log removal of viruses, *Cryptosporidium*, and *Giardia* per month of aquifer travel time.
- Evaluate the attenuation and removal of organic contaminants through SAT, focusing on CECs and TOC and DOC.
- Evaluate the production of DBPs, including trihalomethanes (THMs), haloacetic acids (HAAs), and N-Nitrosodimethylamine (NDMA), as a result of free chlorine or monochloramine injection upstream of the SWIFTRC injection well (TW-1), evaluate the dissipation of free or combined chlorine residual in the soil column, and evaluate the removal of DBPs through SAT.
- Evaluate the attenuation, transformation, and removal of nitrogen species by SAT

Two different travel times were considered as part of Phase 1. One set of two soil columns were used to simulate the monitoring well (MW-SAT) that will be located at the SWIFTRC at a distance of approximately 50 feet from the injection well (TW 1), and these columns are referred to here as the "50-foot" columns. A travel time of 3.2 days was simulated in these columns, based on that estimated using the approach described below:

- Injection at TW 1 was simulated at 1.0 mgd with the flow split between the Lower Potomac aquifer (LPA), Middle Potomac aquifer (MPA), and Upper Potomac aquifer (UPA) based on the expected transmissivities and hydraulic conductivities of each aquifer zone. The flow directed to the UPA was estimated to be 0.42 mgd.
- This flow rate was then used with the screen length in the UPA and an estimated effective porosity
  of 0.35 to determine the radial velocity of the injected water as a function of distance from TW 1
  using a Lagrangian calculation approach.
- This gives 3.2 days travel time in the UPA. Using the same porosity, the soil columns were designed with a feed flow rate of 13 mL/min, a diameter of 12 inches, and soil depth of 7.5 feet.

The other set of two soil columns were used to simulate a somewhat arbitrary travel time of 1 month, and these columns are referred to here as the "1-month" columns. The target feed flow rate was estimated at 2.2 mL/min using the expected aquifer porosity for soil columns with a diameter of 12 inches and a soil depth of 12 feet.

The feed for one of each set of soil columns will be amended with free chlorine and the other with preformed monochloramine, followed by 5 minutes of contact time in the feed tubing, and injection in to the bottom of the columns. The objective is to simulate the two different conditions possible at the SWIFTRC and DBP formation and removal per the objective described above.

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The soil columns were filled with washed aquifer sand material, compacted to remove entrapped air, and then flushed with pilot effluent per protocols developed as part of similar projects. A tracer study was conducted to confirm travel time and tracer dispersion in the column. The columns have been in continuous operation for at least four months. Pathogen removal was assessed by amending the column feed with MS2 coliphage, non-pathogenic *E.coli* K12, and fluorescent microspheres simulating *Cryptosporidium* oospores. Sampling and analysis will be initiated per the plan detailed below to consider nitrogen species, TOC and DOC, CECs, and DBPs. The sampling schedule will be conducted in such a way that the column influent and effluent samples can be compared in the context of the measured travel time for each set of soil columns.

It is important to recognize that for these soil columns, it is not possible to assess hydraulics limitations due to clay mineral fragmentation. This is because the soil used in the column represents well-washed aquifer sands removed from the PAS during monitoring well installation. This washing step was designed to remove well drilling mud contamination, and this effectively removed most of the clay material. Furthermore, it is not possible to extract a truly representative core from the aquifer that would be of a size sufficient to simulate the travel times being considered here.

The dissolved oxygen (DO) concentration of the pilot effluent should be in the range of 15 to 20 milligrams per liter (mg/L) as a result of ozonating water in the upstream treatment process. Efforts were made to prevent contact of the feed water with the atmosphere to prevent stripping of oxygen, and DO probes were mounted in the top of the columns to attempt to measure the DO at the soil column effluent. Temperature is another important parameter for the soil column study. In order to track any effect of temperature on the fate and transport of contaminants or pathogens, DO probes with the ability to take temperature measurements were installed to monitor the temperature of the effluent from the columns. The slow rate of the flow of influent into the columns allowed the water to reach ambient temperature. Therefore, the ambient temperature was monitored through an online temperature sensor which was placed onto the soil column frames. The probes were connected to a programmable logic controller (PLC) to allow for continuous logging of temperature measurements.

Redox conditions were not actively controlled in the soil columns, but DO transport through the columns will be known, and oxidation-reduction potential (ORP) measurements will also be made. This is an important consideration for removal and degradation of organics. However, given the removal of clays from the soil column material, the inability to completely control redox, and the lack of interaction with native groundwater, another limitation of the soil column study is that it is not possible to evaluate the oxidation of reduced iron and mobilization of metals such as manganese and arsenic. As indicated above, this must be evaluated using the SWIFTRC network of monitoring wells.

It is likely that Phase II soil column work will consider travel times in excess of 1 month, perhaps up to as long as 18 months. Phase II is described in Section 3.6 below.

### 3.2 Phase I - Experimental Setup

The system is housed in the SWIFT pilot facility at the York River Treatment Plant at room temperature. The construction of Phase 1 soil columns for HRSD's SWIFT Pilot Program consists of two travel time intervals. As shown on Figure 3.1 two columns were constructed to represent the travel time of replenished water to the monitoring well 50-feet (MW-SAT) from the injection well head (TW-1), and two other columns to represent 1 month (30 days) of aquifer travel time after the water is added to the aquifer. The columns run in parallel to allow for replication of experiments in real time, however one column for each travel time will be fed with free chlorine and one will be fed with monochloramine to simulate the 5-min contact time chlorine contact pipe associated with the SWIFTRC (as described above)

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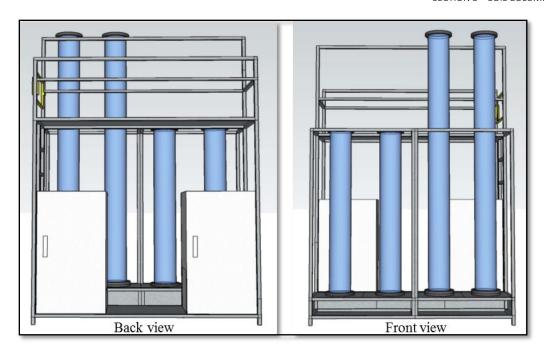


Figure 3-1. SAT Columns Schematic

The diameter of all four columns is 1 foot. The columns that represent 50 feet of the travel from the well head to the monitoring well are 8-feet in length, filled with 7.5-feet of soil. The columns that represent 1-month of aquifer travel time are 13 feet in length, filled with 12 feet of soil. Influent are stored in a refrigerator in a 7-gallon container. The 50-foot columns and the 1-month columns are being fed with parallel precision peristaltic pumps, an upflow configuration at a rate of 13 mL/min and 2.2 mL/min, respectively.

DO probes were installed at the top of each column in a flange face that allow for real time DO measurement. Effluent travels through the top of the column, through a fitting out of the flange face into a sample refrigerator. The columns were constructed from 12-inch schedule-40 clear PVC. Clear PVC was used to ensure all air bubbles were removed from the column during soil filling and flushing. Upon the start of the sampling campaigns, the columns were covered with thick polyethylene plastic with the thickness of ... to prevent light penetration that might encourage algal growth or photolytic chemical transformations not representative in the aquifer. The soil used to fill the columns was washed sand from the Potomac aquifer that was removed during the drilling and construction of the monitoring wells at the SWIFTRC.

## 3.3 Phase I – Soil Column Operation

The SAT columns are being operated on a continuous basis. As shown on Figure 3-2 highly treated water coming from the UVD system is the feed for the SAT columns (SWIFT Pilot Effluent).

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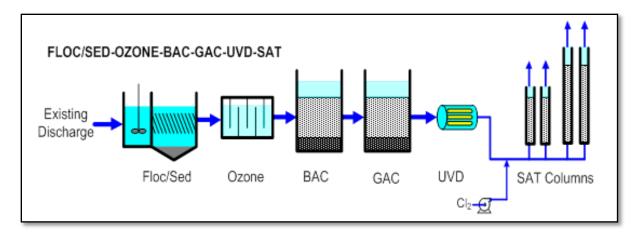


Figure 3-2 SWIFT Pilot Treatment Process

For the 50-feet design SAT column, a total daily volume of 18 L is necessary to continuously feed 13 mL/min, and for the 1-month design SAT column, a total daily volume of 3.1 L is necessary to continuously feed 2.2 mL/min. The UVD system does not operate continuously due to flowrate restrictions, so feeding of the soil columns is accomplished by filling on a daily basis four 7-gallon containers with UVD effluent. As shown on Figure 3-3, the containers with UVD effluent are stored in a refrigerator. Additionally, there are four 5-gallon cubitainers collecting the effluent from each SAT column. All tubing related with the SAT system is replaced at a regular time interval or in the event that biofilm appears. The pumps are calibrated at least twice per week.

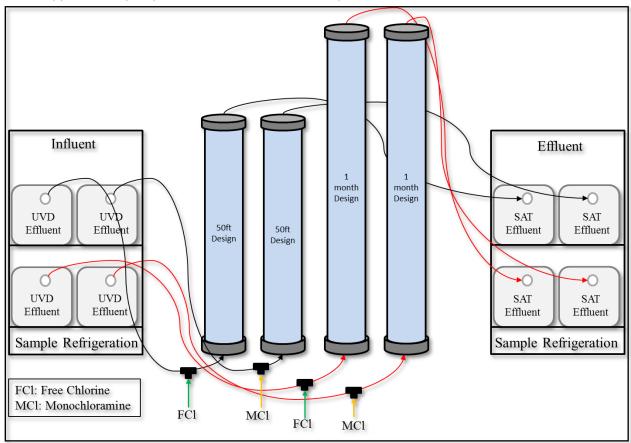


Figure 3-3. SAT Columns Sampling Configuration

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The SWIFTRC was designed to have the capability to either feed free chlorine or monochloramines before recharging the aquifer to avoid biofouling in the well and to achieve virus disinfection credit when using free chlorine and the 5-minute chlorine contact pipeline associated with the SWIFTRC. For Phase 1 soil column work, these two different approaches will be tested to better understand the potential of forming DBPs in the aquifer, followed by removal of DBPs through SAT. Free chlorine or monochloramines will be injected into the feed tubing using precision peristaltic pumps to achieve doses expected at the SWIFTRC and a total of 6-minutes of contact time in the tubing prior to entry of the flow into the soil columns.

Once all SAT columns were ready for operation, soil collected from the Potomac aquifer was washed and sieved to remove material larger than ~4 mm before loading it into the columns. Sieving the soil was done to minimize short-circuiting and flow distortions caused by large debris. As the soil was added to the column, pilot effluent from the SWIFT pilot was added to saturate the soil, and the columns were tapped with a rubber mallet to improve compaction and release trapped air. Sieve analysis was also conducted on the washed sand to determine the media size distribution. The sand was sent to ECS Mid-Atlantic for three sets of sieve analyses.

After obtaining the desired level of soil in the columns, the flushing period started. Flushing was accomplished by pumping SWIFT pilot effluent into the bottom of each column at a higher flowrate (higher than the designed flowrate). Flushing details have been described below:

- Design #1 (50-feet) was run at 3X the designed flow rate for more than 20 bed volumes.
- Design #2 (1-month) was run at 3X the designed flow rate for more than 3 bed volumes.

After the flushing period was completed; a tracer study was performed on each column using an input of sodium chloride continuously over the duration of a few days to confirm the retention time within the columns. Prior to the introduction of tracer, the background concentration of chloride in the influent was analyzed to determine the amount of chloride to be added. During the tracer study, the SAT columns were operated at the design feed flow rates. Effluent samples were collected for chloride analysis ahead of, during, and after the expected passing of the chloride concentration front for each of the four columns, until the chloride concentration decreased to within 10% of the background value. The tracer test was conducted first on the 50-foot columns. The tracer data was used to assess the actual travel time and the chloride dispersion coefficient by fitting the data to a 1D conservative transport model equation. The dispersion coefficient and effective porosity observed were then used as a guide to conduct tracer test on the 1 month columns. The feed flow rates were adjusted based on the results from tracer testing.

Once flow conditions were set and SAT columns were fully operational, microbial challenge testing began. The pathogen and pathogen indicator concentrations in the SWIFT Pilot effluent are consistently below detection, and so challenge testing was the only viable method for evaluating pathogen removal by SAT. MS2 coliphage, non-pathogenic E.coli strain and fluorescent microspheres (substituting inactivated *Cryptosporidium* oospores) were used to spike the feed of the SAT columns. Fluorescent microspheres are similar to the *Cryptosporidium* oocysts in molecular size and are less adsorptive. These microspheres provide advantages over inactivated (irradiated) *Cryptosporidium* oocysts in SAT studies. In demonstrating system performance fluorescent microspheres provide conservative log removals over oocysts. While oocysts (negative surface charge) are removed via physical mechanisms and physicochemical filtration, comparable sized microspheres will be removed by physical mechanisms only. Fluorescent microspheres used by HRSD do not have a net surface charge. Inactivated *Cryptosporidium* oocysts could potentially have different surface charge characteristics from live oocysts depending on the inactivation process which may affect removal as well. So while inactivated oocysts may seem ideal from a health risk perspective, high cost make them impractical. Finally commercial fluorescent microspheres allow high enough concentrations to measure significant removal over several

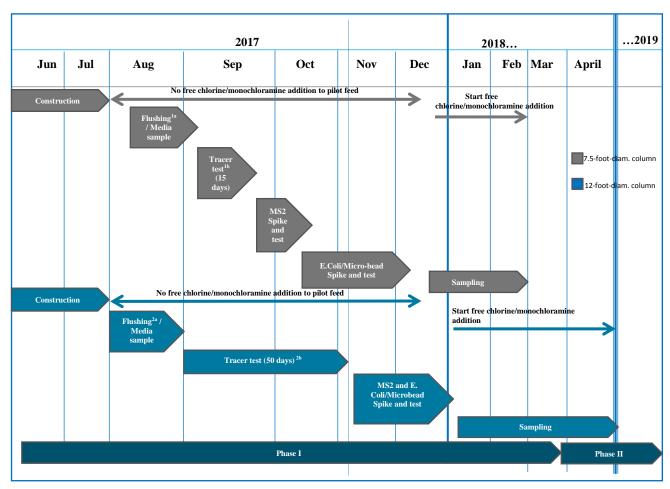
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orders of magnitude. Oocysts are naturally lower in concentration than other pathogens (bacterial and viral) so achieving a desirable seed stock concentration to quantify SAT removal was not practical.

Sampling and analysis will be initiated per the plan described below.

### 3.4 Phase I – Schedule

The SAT columns will be operated in 2 different phases. Phase 1 is taking place at the SWIFT Pilot located at the HRSD York River Treatment Plant, and Phase 2 will take place at the SWIFTRC located at the Nansemond Treatment Plant. The Phase I project schedule is shown in Figure 3-4.



<sup>&</sup>lt;sup>1a</sup> Flushing more than 20 pore volumes at 3X flow rate. Anticipated duration 21 days; <sup>1b</sup> Tracer injected continuously for 4 days. Testing for tracer conducted for 15 days from the point of tracer injection.

Figure 3-4. Schedule of the SAT Columns study

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<sup>&</sup>lt;sup>2a</sup> Flushing more than 3 pore volumes at 3X flow rate. Anticipated duration 30 days; <sup>2b</sup> Tracer injected continuously for 10 days. Testing for tracer conducted for 50 days from the point of tracer injection.

Table 3-1. Duration, flow rates and pore volumes required for tests in 50-foot columns 50-foot Column (Total Volume = 0.167 cubic meter [m³]); Estimated Pore Volume = 0.058 m³ = 58,371 milliliters [mL])

| Experimental Phase                 | Duration (d) | Q (mL/min) | Pore Volumes | Daily Solution<br>Volume (mL) |
|------------------------------------|--------------|------------|--------------|-------------------------------|
| Flushing                           | 22           | 39         | 21.1         | 56,160                        |
| Tracer Injection Duration          | 4            | 13         | 3.1          | 18,720                        |
| Tracer Test Sampling               | 15           | 13         | 4.7          | 18,720                        |
| MS2 Injection Duration             | 5            | 13         | 1.6          | 18,720                        |
| MS2 Sampling                       | 20           | 13         | 9.4          | 18,720                        |
| E.coli and Micro-bead<br>Injection | 5            | 13         | 1.6          | 18,720                        |
| E.coli and Micro-bead<br>Sampling  | 20           | 13         | 6.2          | 18,720                        |
| Monitoring                         | 135          | 13         | 42.2         | 18,720                        |

Table 3-2. Duration, flow rates and pore volumes required for tests in 1-month columns 1-month Column (Total Volume =  $0.267 \text{ m}^3$ ; Estimated Pore Volume =  $0.093 \text{ m}^3 = 93,394 \text{ mL}$ )

| Experimental Phase                            | Duration<br>(d) | Q (mL/min) | Pore Volumes | Daily Solution<br>Volume (mL) |
|---|-----------------|------------|--------------|-------------------------------|
| Flushing                                      | 40              | 6.6        | 3.1          | 9,504                         |
| Tracer Injection Duration                     | 10              | 2.2        | 0.33         | 3168                          |
| Tracer Test                                   | 50              | 2.2        | 1.7          | 3,168                         |
| MS2/E. coli and Micro-bead Spike              | 30              | 2.2        | 3.6          | 3,168                         |
| MS2/ <i>E.coli</i> and Micro-bead<br>Sampling | 50              | 2.2        | 1.7          | 3168                          |
| Monitoring                                    | 90              | 2.2        | 3.0          | 3,168                         |

## 3.5 Phase I – Sampling and Analysis Plan

After the pathogen tests, sampling will begin in the replicate 50-foot soil columns, sampling the column feed for the parameters identified in Table 3-1. Sampling of the effluent from the soil columns for the same parameters will occur based on the anticipated travel time through the columns as identified with the tracer testing, such that the influent and effluent data are comparable. Sampling will occur at the frequency and duration specified in the table and is anticipated to occur in mid-December of 2017 to February of 2018. Sampling for emerging contaminants will begin toward the latter half of the sampling campaign to allow acclimation and accumulation of biomass within the column.

Similarly, sampling for the 1 month replicate columns will occur in the feed with time delayed sampling at the outlet based on anticipated travel time. Refer to Table 3-1 for sampling frequency and duration.

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Dissolved oxygen probes will be installed near the outlets of each set of column to collect continuous DO measurements. The probes will also record temperature readings of the effluents. Temperature sensor will be installed on the soil column frame to monitor ambient temperature.

Samples of the column media will be submitted to Virginia Tech for solid phase TOC analysis and a microbial community analysis. In order to account for the naturally occurring, temporal changes in the microbial community, two control reactors were set up (Figure 3-5); each one receiving pilot feed without free chlorine/monochloramine for the entirety of the study. The two reactors are being run in series, with the first reactor (2 inches in diameter and 3.5 feet high) simulating a travel time of 3.2 days and the second reactor (4 inches in diameter and 7 feet 8 inches high) simulating a travel time of 30 days. The plan was to take triplicate samples of the sand at three different stages. Media samples have been taken from the washed and sieved sand pile before placing in the columns and then from the column tops and the control columns after flushing. Media samples will finally be taken from different heights of the columns and the controls at the end of Phase I. Sterile centrifuge tubes were used for sampling and collection. Each sample was no less than 5 g and the amount sampled was kept consistent in mass volume across samples. Furthermore, a core sample from the aquifer will also be extracted for an overall comparison of the microbial communities.

Table 3-3. Duration, flow rates and pore volumes required for tests in control reactors

| Experimental Phase | Duration (days) | Q (mL/min) | Pore volume (control for 50 feet) | Pore volume (control for 1 month) |
|--------------------|-----------------|------------|-----------------------------------|-----------------------------------|
| Flushing           | 40              | 0.49       | 12.1                              | 1.38                              |
| Monitoring         | 170             | 0.16       | 51.5                              | 5.9                               |

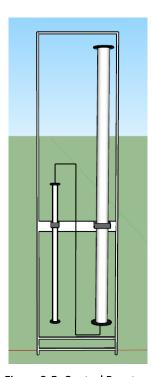


Figure 3-5: Control Reactors

As described above, considerable pathogen removal is anticipated through SAT. In order to confirm the pathogen removal credit available through SAT, MS2, *E.coli* and *Cryptosporidium* challenge tests were conducted using the 50-foot and 1-month travel time columns. For this purpose, the column feeds were spiked with MS2 sufficient to demonstrate > 6-log virus removal, *E.coli* sufficient to demonstrate > 5 log removal and fluorescent microspheres (in lieu of *Cryptosporidium* oocysts) sufficient to demonstrate >

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5-log removal. *E. coli K-12* is the specific strain that was used for the challenge tests. E. coli and fluorescent microspheres were injected in the columns simultaneously.

MS2 were injected in both the 50-foot columns for 5 days followed by injection of microspheres and E. coli. For the 1 month columns, MS2 were injected in one of the columns while E. coli along with microspheres will be injected in another for the duration of 30 days.

Table 3-4. Concentrations of tracer, MS2 and microbeads

|                   | 50 feet                       | 1 month                       |
|-------------------|-------------------------------|-------------------------------|
| Tracer (Chloride) | 500 mg/L as Cl <sup>-</sup>   | 500 mg/L as Cl <sup>-</sup>   |
| MS2               | 10 <sup>7</sup> pfu/mL        | 10 <sup>7</sup> pfu/mL        |
| Microbeads        | 1.24*10 <sup>5</sup> count/mL | 1.35*10 <sup>5</sup> count/mL |
| E. coli           | 10 <sup>6</sup> MPN/mL        | 10 <sup>6</sup> MPN/mL        |

Note:

Cl = chloride

MPN = most probable number

pfu = plaque forming unit

Concentration data for various constituents will be used to analyze loss coefficients to a 1D reactive transport model. The equation will include sorption/desorption and biodegradation and decay terms. Microbial data will be modeled using a particle-transport model that accounts for sorption, straining, and growth/decay dynamics.

Table 3-5. Sampling parameters for replicate soil columns. Sampling to occur in the feed and the effluent.

| Parameter                    | Method               | Sampling<br>Frequency | Duration | Anticipated<br>Timeframe |  |
|------------------------------|----------------------|-----------------------|----------|--------------------------|--|
|                              | 50-foot column       |                       |          |                          |  |
| Ammonia (NH <sub>3</sub> )   | Lachat 10-107-06-1-C | 3x/week               | 2 months | Dec-Feb                  |  |
| TKN                          | Lachat 10-107-06-2-I | 3x/week               | 2 months | Dec-Feb                  |  |
| Nitrate (NO <sub>3</sub> )   | Calculation          | 3x/week               | 2 months | Dec-Feb                  |  |
| Nitrite (NO <sub>2</sub> )   | Lachat 10-107-04-1-C | 3x/week               | 2 months | Dec-Feb                  |  |
| Nitrous Oxide                |                      | 3x/week               | 2 months | Dec-Feb                  |  |
| Dissolved Organic Carbon     | SM 5310 B-2011       | 3x/week               | 2 months | Dec-Feb                  |  |
| TOC                          | SM 5310 B-2011       | 3x/week               | 2 months | Dec-Feb                  |  |
| Disinfection Byproducts      |                      |                       |          |                          |  |
| (HAA5)                       | EPA 552.2            | 3x/week               | 2 months | Dec-Feb                  |  |
| Total Trihalomethanes        | EPA 624              | 3x/week               | 2 months | Dec-Feb                  |  |
| Orthophosphate               | Lachat 10-115-01-1-A | 3x/week               | 2 months | Dec-Feb                  |  |
| Total Phosphorus             | Lachat 10-115-01-1-E | 3x/week               | 2 months | Dec-Feb                  |  |
| Indicator CECs <sup>1</sup>  | Varied               | 2x/week               | 3 weeks  | Dec-Feb                  |  |
| Additional CECs <sup>2</sup> | Varied               | 1x/week               | 3 weeks  | Dec-Feb                  |  |
| 1 month column               |                      |                       |          |                          |  |
| NH <sub>3</sub>              | Lachat 10-107-06-1-C | 1x/week               | 3 months | Jan- Mar                 |  |
| TKN                          | Lachat 10-107-06-2-I | 1x/week               | 3 months | Jan- Mar                 |  |
| Nitrate                      | Calculation          | 1x/week               | 3 months | Jan- Mar                 |  |
| $NO_2$                       | Lachat 10-107-04-1-C | 1x/week               | 3 months | Jan- Mar                 |  |
| Dissolved Organic Carbon     | SM 5310 B-2011       | 1x/week               | 3 months | Jan- Mar                 |  |

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| Parameter                    | Method               | Sampling<br>Frequency | Duration | Anticipated<br>Timeframe |
|------------------------------|----------------------|-----------------------|----------|--------------------------|
| TOC                          | SM 5310 B-2011       | 1x/week               | 3 months | Jan- Mar                 |
| Disinfection Byproducts      |                      |                       |          |                          |
| (HAA5)                       | EPA 552.2            | 1x/week               | 3 months | Jan- Mar                 |
| Total Trihalomethanes        | EPA 624              | 1x/week               | 3 months | Jan- Mar                 |
| Orthophosphate               | Lachat 10-115-01-1-A | 1x/week               | 3 months | Jan- Mar                 |
| Total Phosphorus             | Lachat 10-115-01-1-E | 1x/week               | 3 months | Jan- Mar                 |
| Indicator CECs <sup>1</sup>  | Varied               | 2x/month              | 2 months | Jan-Feb                  |
| Additional CECs <sup>2</sup> | Varied               | 1x/month              | 2 months | Jan-Feb                  |

<sup>&</sup>lt;sup>1</sup> Refer to Table 7-2 in Attachment B for detailed list

Notes:

EPA = U.S. Environmental Protection Agency

 ${\sf HAA5} = {\sf haloacetic\ acids\ (monochloroacetic,\ dichloroacetic,\ trichloroacetic,\ dichloroacetic,\ d$ 

monobromoacetic, and dibromoacetic)

TKN = total kjeldahl nitrogen

Table 3-6. *Identification of individual CECs included in the soil column sampling and analysis plan.* 

| Additional CECs                              | Rationale for Monitoring |  |  |
|--|--------------------------|--|--|
| Cyanotoxins                                  |                          |  |  |
| Total microcystin                            | CCL4                     |  |  |
| Anatoxin-a                                   | CCL3/CCL4                |  |  |
| Cylindrospermopsin                           | CCL3/CCL4                |  |  |
| Microcystin-LR                               | CCL3/CCL4                |  |  |
| Disinfection Byproduc                        | ts                       |  |  |
| Chlorate                                     | CCL4                     |  |  |
| Bromochloroacetic acid                       | UCMR4                    |  |  |
| Bromodichloroacetic acid                     | UCMR4                    |  |  |
| Dibromochloroacetic acid                     | UCMR4                    |  |  |
| Tribromoacetic acid                          | UCMR4                    |  |  |
| Flame Retardants                             |                          |  |  |
| BDE-100                                      | Chemical of interest     |  |  |
| BDE-153                                      | Chemical of interest     |  |  |
| BDE-154                                      | Chemical of interest     |  |  |
| BDE-183                                      | Chemical of interest     |  |  |
| BDE-209                                      | Chemical of interest     |  |  |
| BDE-28                                       | Chemical of interest     |  |  |
| BDE-47                                       | Chemical of interest     |  |  |
| BDE-99                                       | Chemical of interest     |  |  |
| Bromochloromethane                           | CCL3/CCL4/UCMR3          |  |  |
| Bromomethane                                 | CCL3/CCL4/UCMR3          |  |  |
| Tris(2-chloroethyl) phosphate (TCPP)         | Chemical of interest     |  |  |
| Tris(1,3-dichloro-2-propyl)phosphate (TDCPP) | Chemical of interest     |  |  |

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<sup>&</sup>lt;sup>2</sup> Identified in Table 3-6 of this Attachment

| Additional CECs                 | Rationale for Monitoring |
|---------------------------------|--------------------------|
| Hormone, Natural or Syn         | thetic                   |
| 16-α-hydroxyestradiol (estriol) | CCL3/CCL4/UCMR3          |
| 17-α-ethynylestradiol           | CCL3/CCL4/UCMR3          |
| 17-β-estradiol                  | CCL3/CCL4/UCMR3          |
| 4-androstene -3,17-dione        | UCMR3                    |
| Androstenedione                 | Chemical of interest     |
| Equilin                         | CCL3/CCL4/UCMR3          |
| Estradiol                       | Chemical of interest     |
| Estriol                         | Chemical of interest     |
| Norethindrone                   | CCL3/CCL4                |
| Progesterone                    | Chemical of interest     |
| Testosterone                    | UCMR3                    |
| Pharmaceutical/Personal Care/Fo | ood derivatives          |
| Theobromine                     | Chemical of interest     |
| 1,7-Dimethylxanthine            | Chemical of interest     |
| Acesulfame-K                    | Chemical of interest     |
| Butylparaben                    | Chemical of interest     |
| Caffeine                        | Chemical of interest     |
| Ethylparaben                    | Chemical of interest     |
| Isobutylparaben                 | Chemical of interest     |
| Methylparaben                   | Chemical of interest     |
| Musk Ketone                     | Chemical of interest     |
| Propylparaben                   | Chemical of interest     |
| Triclocarban (TCC)              | Chemical of interest     |
| Acetaminophen                   | Chemical of interest     |
| Albuterol                       | Chemical of interest     |
| Amoxicillin                     | Chemical of interest     |
| Atenolol                        | Chemical of interest     |
| Azithromycin                    | Chemical of interest     |
| Bendroflumethiazide             | Chemical of interest     |
| Bezafibrate                     | Chemical of interest     |
| Butalbital                      | Chemical of interest     |
| Carbadox                        | Chemical of interest     |
| Carisoprodol                    | Chemical of interest     |
| Chloramphenicol                 | Chemical of interest     |
| Cimetidine                      | Chemical of interest     |
| Clofibric Acid                  | Chemical of interest     |
| Dehydronifedipine               | Chemical of interest     |
| Diazepam                        | Chemical of interest     |
| Diclofenac                      | Chemical of interest     |
| Dilantin                        | Chemical of interest     |
| Diltiazem                       | Chemical of interest     |
| Erythromycin                    | CCL3/CCL4                |
| Flumequine                      | Chemical of interest     |

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| Additional CECs                      | Rationale for Monitoring |
|--------------------------------------|--------------------------|
| Fluoxetine                           | Chemical of interest     |
| Galaxolide                           | Chemical of interest     |
| Gemfibrozil                          | Chemical of interest     |
| Ibuprofen                            | Chemical of interest     |
| lohexol                              | Chemical of interest     |
| lopromide                            | Chemical of interest     |
| Ketoprofen                           | Chemical of interest     |
| Ketorolac                            | Chemical of interest     |
| Lidocaine                            | Chemical of interest     |
| Lincomycin                           | Chemical of interest     |
| Linuron                              | Chemical of interest     |
| Lopressor                            | Chemical of interest     |
| Meclofenamic Acid                    | Chemical of interest     |
| Naproxen                             | Chemical of interest     |
| Nifedipine                           | Chemical of interest     |
| Oxolinic Acid                        | Chemical of interest     |
| Pentoxifylline                       | Chemical of interest     |
| Phenazone                            | Chemical of interest     |
| Propazine                            | Chemical of interest     |
| Quinoline                            | CCL3/CCL4/UCMR4          |
| Sulfachloropyridazine                | Chemical of interest     |
| Sulfadiazine                         | Chemical of interest     |
| Sulfadimethoxine                     | Chemical of interest     |
| Sulfamerazine                        | Chemical of interest     |
| Sulfamethazine                       | Chemical of interest     |
| Sulfamethizole                       | Chemical of interest     |
| Sulfamethoxazole                     | Chemical of interest     |
| Sulfathiazole                        | Chemical of interest     |
| Theophylline                         | Chemical of interest     |
| Thiabendazole                        | Chemical of interest     |
| Trimethoprim                         | Chemical of interest     |
| Warfarin                             | Chemical of interest     |
| Perfluorinated Compou                | ınds                     |
| Perfluorobutanesulfonic Acid (PFBS)  | UCMR3                    |
| Perfluoroheptanoic Acid (PFHpA)      | UCMR3                    |
| Perfluorohexanesulfonic Acid (PFHxS) | UCMR3                    |
| Perfluoroocnanoic Acid (PFNA)        | UCMR3                    |
| Perfluorooctanesulfonic Acid (PFOS)  | CCL3/CCL4/UCMR3          |
| Perfluorooctanoic Acid (PFOA)        | CCL3/CCL4/UCMR3          |
| Pesticides                           |                          |
| 3-Hydroxycarbofuran                  | CCL3/CCL4                |
| Bifenthrin                           | Chemical of interest     |
| Bromacil                             | Chemical of interest     |
| Chloridazon                          | Chemical of interest     |

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| Additional CECs                   | Rationale for Monitoring |
|-----------------------------------|--------------------------|
| Chlorotoluron                     | Chemical of interest     |
| Chlorpyrifos                      | CCL4                     |
| cis-Permethrin                    | UCMR4                    |
| Cyanazine                         | Chemical of interest     |
| Diaminochloro-atrazine (DACT)     | Chemical of interest     |
| Desethyl-atrazine (DEA)           | Chemical of interest     |
| Desisopropyl-atrazine (DIA)       | Chemical of interest     |
| Dimethoate                        | CCL3                     |
| Disulfoton                        | CCL3                     |
| Diuron                            | CCL3/CCL4                |
| Fenitrothion                      | Chemical of interest     |
| Fipronil                          | Chemical of interest     |
| Isoproturon                       | Chemical of interest     |
| Kepone                            | Chemical of interest     |
| Metazachlor                       | Chemical of interest     |
| Sulfometuron, methyl              | Chemical of interest     |
| Permethrins, Total (cis-, trans-) | UCMR4                    |
| Picloram                          | Chemical of interest     |
| Tributyltin (nanograms per liter) | Chemical of interest     |
| Semivolatile Organic              | 5                        |
| 4-nonylphenol - semi quantitative | CCL4                     |
| 4-tert-octylphenol                | CCL4                     |
| Aniline                           | CCL3/CCL4                |
| Bisphenol A                       | Chemical of interest     |
| Nitrobenzene                      | CCL3/CCL4                |
| n-Nitrosodiethylamine             | CCL3/CCL4                |
| n-Nitrosodi-n-propylamine         | CCL3/CCL4                |
| n-Nitrosodiphenylamine            | CCL3/CCL4                |
| n-Nitrosopyrrolidine              | CCL3/CCL4                |
| Nonylphenol                       | CCL4                     |
| Propylbenzene                     | CCL3/CCL4                |
| Volatile Organics                 |                          |
| 1,1,1,2-Tetrachloroethane         | CCL3/CCL4                |
| 1,1-Dichloroethane                | CCL3/CCL4/UCMR3          |
| 1,2,3-Trichloropropane            | CCL3/CCL4/UCMR3          |
| 1,3-butadiene                     | CCL3/CCL4/UCMR3          |
| Acrolein                          | CCL3/CCL4                |
| Chlorodifluoromethane (HCFC-22)   | UCMR3                    |
| Chloromethane                     | CCL3/CCL4/UCMR3          |
| Formaldehyde                      | CCL3/CCL4                |
| Hexane                            | CCL3/CCL4                |
| Methanol                          | CCL3/CCL4                |
| Methyl tert-Butyl Ether           | CCL3/CCL4                |
| sec-Butylbenzene                  | CCL3/CCL4                |
|                                   | ,                        |

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| Additional CECs | Rationale for Monitoring |
|-----------------|--------------------------|
| Other           |                          |
| Bromide         | UCMR4                    |
| Cobalt          | CCL4                     |
| Germanium       | CCL4/UCMR4               |
| Molybdenum      | CCL4                     |
| Tellurium       | CCL4                     |
| Vanadium        | CCL4                     |

BDE = brominated diphenyl ether

CCL = Candidate Contaminant List

UCMR = Unregulated Contaminant Monitoring Rule

# 3.6 Phase II Planning – SWIFTRC Soil Column Testing

The SWIFTRC is expected to commence operations in late January 2018. During initial operation, SWIFT Water will be recirculated to Nansemond Treatment Plant and ultimately discharged through the plant outfall. Once sufficient operational time has elapsed to allow HRSD to confirm the function of the SWIFTRC mechanical systems, instrumentation and controls, and SWIFT Water quality, MAR will commence using the SWIFTRC Recharge Well (TW-1) at flow rates increasing to 1 mgd.

As indicated above, the results of Phase I soil column testing work will be used in an adaptive management approach to guide the development of Phase II. The Phase II SAT testing program will likely be primarily focused on demonstration of fate of SWIFT Water constituents and microbial surrogates for travel times in excess of those investigated during Phase I (i.e. greater than 1 month), but work from Phase I will also be repeated using SWIFT Water from the SWIFTRC. Longer SAT retention times will be monitored in-situ throughout the demonstration period at wells MW-UPA, MW-MPA and MW-LPA, located 400, 450 and 500 feet from TW-1, respectively. For example, injectate water is expected to reach the monitoring wells between 6-12 months after commencement of recharge operations. However, as with Phase I, ex-situ testing in soil columns will allow additional data collection on temporal changes in injectate constituents during SAT and will allow investigation of removal of spiked microbial contaminant surrogates and perhaps also a cocktail of various organic chemicals of interest.

Additional columns will be constructed and installed at the SWIFTRC for the purpose of Phase II SAT testing. The columns will be designed to replicate 6 months of SAT and possibly up to 18 months. Design details, including physical construction and instrumentation, will generally replicate that of the Phase I columns. The sampling and analysis plan outlined for the Phase I columns will largely be followed for Phase II with the exception of sample frequency (preliminary planned to be monthly) and testing duration.

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# Field Scale Testing of SAT at SWIFT Research Center

The facilities associated with MAR activities involved with the field scale SAT at the SWIFTRC include the test recharge well (TW-1) and a multi-aquifer monitoring well (MW-SAT) located 50 feet away from TW-1 (Figure 4-1).



Figure 4-1. Map of TW-1, MW-SAT, MW-UPA, MW-MPA, and MW-LPA at Nansemond WWTP

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In addition to MW-SAT, the SWIFT facility will include three conventional monitoring wells (MW-UPA, MW-MPA, and MW-LPA) lying at distances ranging between 400 and 500 feet from TW-1, each screening multiple sand intervals in the Upper, Middle and Lower zones of the Potomac aquifer, respectively. Because of their distances, travel time in the PAS, and multi-screen construction, sampling results from MW-UPA, MW-MPA, and MW-LPA are not anticipated to influence the SAT studies. However, data from these wells will help characterize the geochemical environment and the transport behavior of some solute in the PAS. Construction and sampling of the conventional monitoring wells is addressed in Attachment C of the UIC Inventory.

SWIFT Water will leave the SWIFTRC and be pumped to TW 1. To discriminate between monitoring the SWIFT advanced water treatment processes and monitoring the aquifer response to MAR, this plan describes water exiting the advanced water treatment facility as "SWIFT Water," and describes water injected into TW-1 as "recharge water." HRSD will possess the capability to measure field chemistry and collect samples of the SWIFT Water, native groundwater, and ultimately recharge water from MW-SAT in the pilot area within the SWIFTRC.

### 4.1 Test Injection and Multi-Aquifer Monitoring Well

This section describes test injection well (TW-1) and MW-SAT and their roles in the SAT field-scale study.

#### 4.1.1 Managed Aquifer Recharge Well

The MAR well (TW 1) extends to 1,410 feet below grade (fbg) and features a 12-inch-diameter carbon steel casing and 380 feet of stainless steel, 0.04-inch slot, continuous wire wrap screen (Figure 4-2).

TW-1 screens the Upper (120 feet), Middle (125 feet), and upper portion of the Lower (135 feet) zones of the PAS. The static water level in TW-1 reflects combined heads from each aquifer and fluctuates around 95 fbg. Water levels measured from the isolated aquifer units during packer testing varied from 95 to 97 fbg.

TW 1 will be equipped with a pressure transducer to measure and record static, injection, and backflushing water levels during MAR operations. HRSD will collect a total of 4 background groundwater samples from TW-1 before starting MAR operations. Because TW-1 screens multiple sand intervals in the PAS, the sample will represent water mixed from multiple zones. HRSD will monitor the quality of the SWIFT Water inside the SWIFTRC, as discussed in Attachment B, SWIFT Research Center SWIFT Water Quality Targets.

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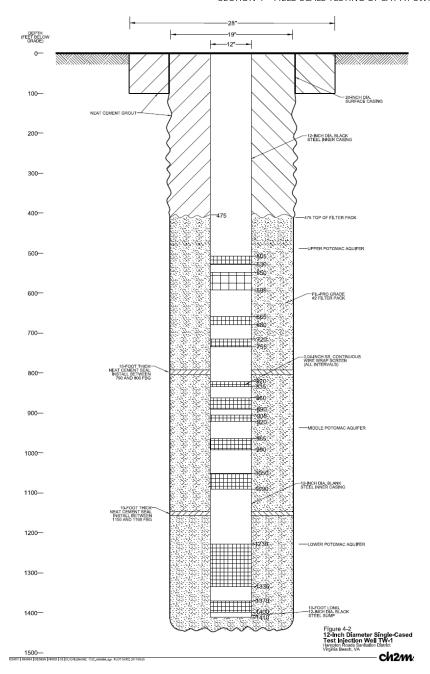


Figure 4-2. 12-inch Diameter Single-Cased Test Injection Well TW-1

### 4.1.2 Multi-Aquifer Monitoring Well (MW-SAT)

#### 4.1.2.1 MW-SAT Well Construction

MW-SAT will lie approximately 50 feet from TW-1 and will support evaluating SAT in the PAS, in response to MAR operations at TW-1. MW-SAT will consist of a 6-inch-diameter carbon steel casing and 380 feet of stainless steel, continuous wire wrap screen extending to 1,410 fbg, the same depth as TW-1. The screen zones in MW-SAT will match the same intervals in TW-1 to the greatest extent practical.

#### 4.1.2.2 MW-SAT FLUTe Sampling System

After installing and developing MW-SAT, a flexible liner, discrete-interval sampling system manufactured by FLUTe (Figure 4-3) will be installed in the casing and screen assembly.

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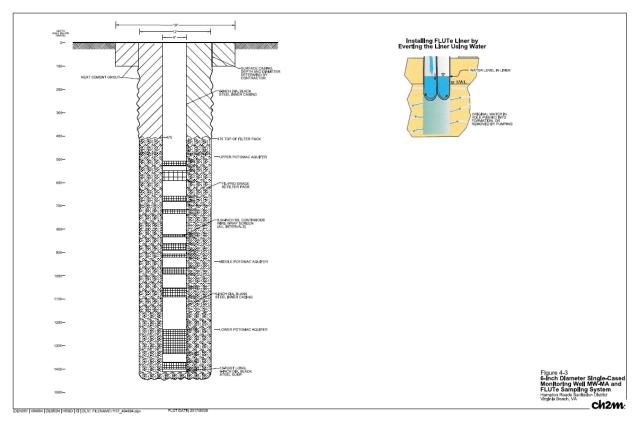


Figure 4-3. 6-inch Diameter Single-Cased Monitoring Well MW-MA and FLUTe Sampling System

The sampling system will consist of eleven sampling ports coinciding with each well screen. Sample tubing extending from each port will run to the ground surface and into the pilot area inside the SWIFTRC where HRSD can control purging, measure field chemistry and collect samples for laboratory analysis from the depth discrete intervals (Figure 4-4 Schematic of FLUTe system).

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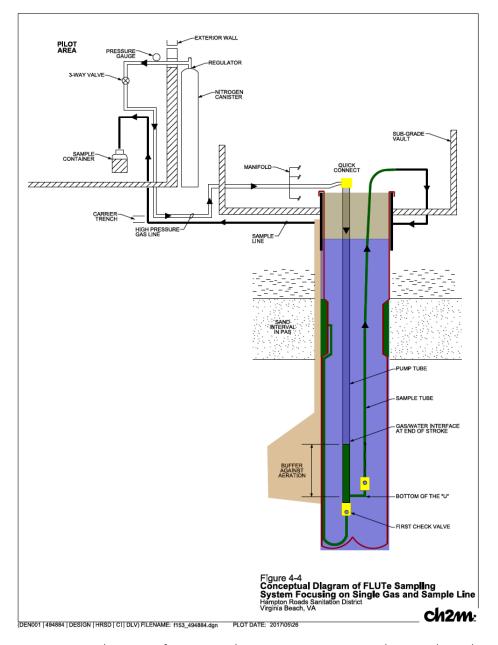


Figure 4-4. Conceptual Diagram of FLUTe Sampling System Focusing on Single Gas and Sample Line

Nitrogen gas from three canisters mounted on the exterior wall of the pilot area will drive sampling and purging. Operators will control purging and sampling from a three-way valve mounted on a panel in the pilot area (Figure 4-5 Panel diagram). To offer greater flexibility in selecting sampling intervals, three manifolds will segregate the gas-feed and sampling tubing by aquifer zone (UPA, MPA, and LPA zones). Recharge water may reach the deeper screens in TW-1 much later than the upper screens, and accordingly may not require the same sampling frequency.

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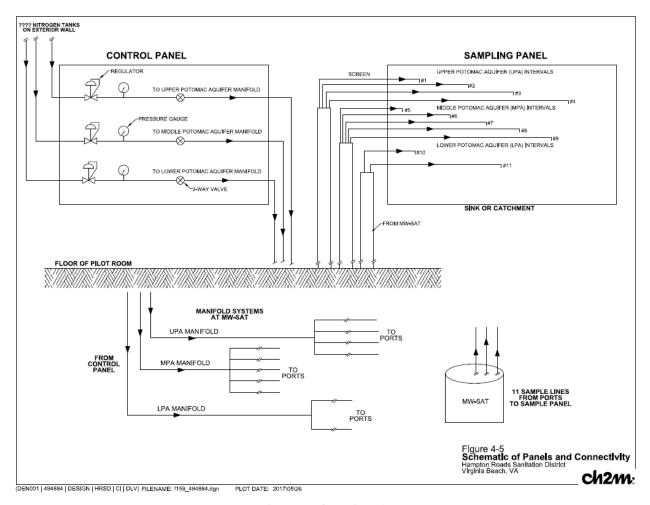


Figure 4-5. Schematic of Panels and Connectivity

MW-SAT will serve multiple, important roles in monitoring the geochemical response to MAR operations in the PAS. First, monitoring at MW-SAT will support characterizing hydrodynamic factors (advection, dispersion, mixing, etc.) influencing solute transport in the PAS, an important consideration for SAT. MW-SAT will also serve as a station to evaluate SAT of selected trace organic compounds including THMs, HAAs, CECs, NDMA, etc., along with the leaching of metals (iron, manganese, aluminum and arsenic) from reactive metal bearing minerals. Pathogens and pathogen indicators will not be monitored at MW-SAT, because it is almost certain that these will already be well below detectable levels in the SWIFTRC SWIFT Water, even with sampling of very large volumes of water. Pathogen removal by SAT must be assessed by challenge testing of the soil columns as described in Section 3.

## 4.2 Field Scale Monitoring Plan

The monitoring plan for evaluating SAT at the field scale includes monitoring SWIFT Water quality, native groundwater, and eventually recharge water migrating in the PAS.

#### 4.2.1 SWIFTRC SWIFT Water

SWIFT Water will continually discharge into a sample sink in the SWIFTRC laboratory and will be continuously monitored by online analyzers for temperature, pH, turbidity, specific conductivity, and DO. Table 9 in Attachment B of the UIC Inventory details the samples and sample frequency identified

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for demonstrating SWIFT Water regulatory compliance, treatment efficacy of trace contaminants, and important recharge water chemistry.

#### 4.2.2 Multi-Aquifer Monitoring Well

MW-SAT, located only 50 feet from TW-1, will serve multiple roles in evaluating field scale SAT during MAR operations at the SWIFTRC including:

- Characterizing hydrodynamic elements (advection, dispersion, mechanical mixing etc.) of solute transport in the PAS system.
- Describing redox conditions at the interface between recharge and native groundwater.
- Helping determine the magnitude of cation exchange between the recharge and aquifer.
- Quantifying the attenuation and treatment/removal of major constituents in the recharge, including DO, nitrate, TKN, phosphorus, orthophosphate, TOC, DOC, chemical oxygen demand, and several others.
- Characterizing SAT of selected trace organic compounds such as THMs, HAAs, CECs, and NDMA.
- Monitoring the leaching of undesirable metals from minerals in the PAS including iron, manganese, and arsenic.

A total of 4 background samples will be collected from MW-SAT prior to recharge operations.

#### 4.2.2.1 Tracer Selection

Tracer selection is discussed in detail in Attachment C, section 2.3.1 of this UIC Inventory. A tracer should be non-reactive between water types and minerals in the aquifer and significantly differing in concentration than native groundwater. Chloride fits these two criteria as a relatively inert ion, differs significantly in concentration between the recharge water (220 mg/L) and groundwater produced from the three aquifers (1,970 to 2,760 mg/L). Specific conductivity is not a direct measurement of the amount of chloride in the water, however, it is easy to measure and can be used to screen for chloride. Both specific conductivity and chloride will be measured and used to identify the recharge water front moving through the monitoring well.

#### 4.2.2.2 Estimated Travel Time

Located only 50 feet away, recharge may reach at least some of the screen intervals in MW-SAT relatively rapidly. If recharge spreads evenly across the eleven screen intervals, totaling 380 feet in length, HRSD will need to recharge 5.8 million gallons before its arrives at MW-SAT (Table 4-1). Dividing the volume by the recharge rate (1 mgd) provides the time (5.8 days) for recharge to arrive at MW-SAT.

Table 4-1. Volumes and times for recharge to reach intervals in MW-SAT

| Monitoring Well     | Recharge Entering Well<br>Screens | Without Dispersion<br>Volume (mgd) | With Dispersion in Sand Aquifer<br>Volume (mgd) |
|---------------------|-----------------------------------|------------------------------------|---|
| MW-SAT <sup>1</sup> | All screens <sup>2</sup>          | 5.8                                | 2.1   |
|                     | Top UPA Only <sup>3</sup>         | 0.4                                | 0.14  |
|                     | UPA Only <sup>4</sup>             | 2.1                                | 0.8   |
|                     | UPA and MPA <sup>5</sup>          | 3.9                                | 1.5   |

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#### Notes:

Because recharge is assumed to spread evenly across the eleven screens, this duration represents the maximum time for recharge to reach MW-SAT. However, several factors can reduce the time for recharge to arrive at a monitoring point, including hydrodynamic dispersion (longitudinal dispersion in the aquifer, recharge channeling along higher permeability pathways, and density segregation. Considering dispersion, it is expected that recharge water will reach MW-SAT after approximately 2 million gallons, taking about 2 days.

During MAR operations, recharge water will more likely exit TW-1 through the uppermost screens and migrate preferentially through the UPA and portions of the MPA, before the LPA. Hypothetically, if all the recharge enters the uppermost screen interval in the UPA, which measures 25 feet in length, water could arrive at MW-SAT after only 0.14 days, if influenced by dispersion, or 0.4 days, if not.

#### 4.2.2.3 Breakthrough Curve

Characterizing the relationship between advection and dispersion in each sand interval screened by MW-SAT using chloride as a tracer will establish a sound basis for evaluating groundwater and solute velocities. The curves will support evaluating the fate of constituents in the PAS other than chloride, including the attenuation of major ions, trace metals, nutrients, and trace organic components undergoing SAT (Figure 4-6). The concentration versus time relation at the monitoring point is often called the breakthrough curve. The geometry of the curve for an individual solute in relation to the tracer's curve can help an analyst interpret the attenuation experienced by the solute. HRSD will compare breakthrough curves for constituents sampled at MW-SAT and from samples exiting the soil columns.

An important factor in reducing transport data and interpreting breakthrough curves will involve knowing the exact linear distance between TW-1 and MW-SAT. Because of the depth of the wells, the distance may differ between shallow and deeper screen intervals, depending on the slope of each wellbore away from true plumb. The exact distances between the screen intervals in TW-1 and MW-SAT will be obtained using gyroscopic surveys conducted in the two wells used to estimate distances based on the slope in each.

#### 4.2.2.4 Water Quality Monitoring

The FLUTe sampling system installed in MW-SAT will consist of polyvinylidene fluoride (PVFD) tubing running from a sample port situated in each well screen to the ground surface. PVFD tubing exhibits the same inert, chemical characteristics as Teflon, but is stronger and less likely to kink during installation or sampling operations. Collection of rinseate samples through PVFD tubing conducted by FLUTe has yielded less than method detection limits for all CEC constituents.

Sampling personnel will employ a nitrogen source to purge the tubing and then withdraw samples from each interval. The use of nitrogen prevents aeration and the alteration of redox in a sample. The pumping system accommodates filling sample bottles quickly and efficiently. The system will also allow attachment of tubing for connection to a flow-through cell to measure important field chemistry constituents (temperature, pH, specific conductivity, ORP, and DO). The 11 sampling tubes will be piped to a sample sink in the piloting area of SWIFTRC.

Groundwater samples will be collected from each of the eleven ports installed in the center of the screen intervals, prior to starting MAR operations. Monitoring the recharge as it first flows past MW-SAT

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<sup>&</sup>lt;sup>1</sup>MW-SAT located 50 feet away from TW-1

<sup>&</sup>lt;sup>2</sup>All screen intervals in TW-1 total 385 feet

<sup>&</sup>lt;sup>3</sup>Top screen in UPA equals 25 feet in length

<sup>&</sup>lt;sup>4</sup>Screen length in UPA equals 140 feet

<sup>&</sup>lt;sup>5</sup>Screen length in UPA and MPA equals 265 feet

in each affected screen interval represents a critical element in discriminating between advective and dispersive transport of a solute in the PAS. Accordingly, samples collected must capture the change in water chemistry as recharge water displaces groundwater in the sand beds of the PAS. This monitoring will require recording specific conductivity measurements, and analyzing chloride in the field using titrators every 12 hours after MAR operations commence at 1 mgd.

Once recharge is detected in a specific interval using specific conductivity and chloride, operators should plan on collecting samples for a more comprehensive suite of analytes at 12 hour intervals in the uppermost screen intervals of MW-SAT. Specific conductivity and chloride measurements collected at 24 hour intervals should continue in deeper screens until measurable changes in water chemistry are encountered. At a minimum, all regulatory limit parameters and performance indicators should be measured on a consistent basis. These parameters are identified in Attachment C of this UIC Inventory.

Once specific conductivity, chloride, calcium, sodium, magnesium, potassium (cations), sulfate, alkalinity (anions), iron, manganese, aluminum (trace metals), nitrate, TKN, nitrite, total phosphorous, orthophosphate as P (nutrients), TOC/DOC, and total dissolved solids (TDS) in samples from a sand interval in MW-SAT equal the concentrations observed in samples from the SWIFT Water, or are stable after 3 samples, operators can reduce the sampling frequency for these constituents, and the interval, to weekly for one month and then monthly, thereafter, with the exception of arsenic. Samples should continue to be collected and analyzed for arsenic on a weekly frequency over the duration of the project.

Daily monitoring for specific conductivity and chloride in deeper screen intervals should continue until concentrations change, indicating the presence of recharge water. Once recharge is detected, operators should collect samples for cations, anions, trace metals, nutrients, TOC/DOC, and TDS analysis on a daily basis. As concentrations of the constituents from the MW equal the concentrations in the SWIFT Water or are stable after three samples, sampling frequency can be reduced to monthly.

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# Evaluating Column and Field Scale Testing Results

Column and field scale testing will produce an enormous amount of field measurements and laboratory analytical data. This section describes some of the techniques HRSD anticipates employing to reduce and then evaluate these data. To achieve uniformity in the evaluation, HRSD will coordinate the analysis of column and field scale testing data. Accordingly, analysis of the column and field scale testing studies are considered together in this section, with narrative differences in analytical approaches between the studies, as appropriate.

Many analytical techniques are described in recent literature on treating data from column and field scale testing studies. This section discusses a few of the more, obvious and rudimentary techniques. HRSD will update this plan, as the analysis grows and deviates from this plan as the data emerges from the study.

## 5.1 SAT Inventory and Breakthrough Tracking

A critical element of the column and field scale testing will involve accounting for parameters that exhibited full or partial breakthrough, and those that did not appear in samples collected during the testing. Many CECs will attenuate during transport through the columns or the PAS over the finite testing duration. Conversely, most of the field chemistry parameters, cations, anions, trace metals, and nutrients should achieve partial to full breakthrough in at least the shallower sample ports of MW-SAT. The number of sand intervals and the changing hydraulic regime in TW-1, as injection heads change or the well clogs will complicate the evaluation of some results, particularly trace organics.

HRSD will need to account for the constituents achieving breakthrough, and those that fail to appear in samples that exit the columns or the sample ports from MW-SAT. In developing the inventory, HRSD can characterize pathogen (soil column only), or organics removal for constituents that did not exit the column or appear in samples collected from the sample ports at MW-SAT.

## 5.2 Characterizing the hydrodynamic signature

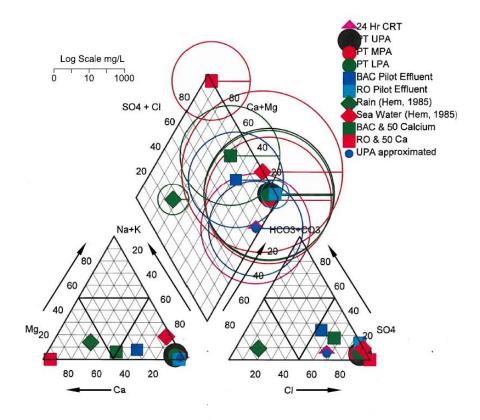
A simple evaluation of breakthrough will involve plotting constituent concentrations (Ct) recovered at a column or sample port divided by the starting concentration (Co) on the Y-axis against time (X-axis). The plot will allow characterizing amount of breakthrough exhibited by the constituent. A partial or complete breakthrough curve should accommodate characterizing how hydrodynamic factors influence the transport of constituent. This evaluation forms an important starting point for describing mechanisms that attenuated the constituent during migration through the column or PAS.

## 5.3 Defining the Geochemical Environment

An important intermediate step in characterizing constituent attenuation entails characterizing the geochemical environment of the columns or discrete sand intervals in the PAS. Geochemical indicators include pH, bulk water chemistry, ionic strength, redox, and TOC/DOC content. Several tools are available to address these factors. An analyst should develop a Piper diagram (Figure 5-1) of the ionic water chemistry, calculate the ionic strength using simple analytical techniques or geochemical modeling, and characterize redox by plotting averaged ORP measurements on an Eh diagram (Figure 5-2). A new analytical method developed by the United States Geological Survey (Jurgens et al., 2009)

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considers common redox indicators beyond ORP (DO, nitrate, manganese, iron, sulfate, sulfide) to describe the redox environment.



#### **LEGEND**

RO - REVERSE OSMOSIS

MU - REVERSE OSMOSIS
BAC - BIOLOGICALLY ACTIVATED CARBON
UPA - UPPER POTOMAC AQUIFER
MPA - MIDDLE POTOMAC AQUIFER
LPA - LOWER POTOMAC AQUIFER
CRT - CONSTANT RATE AQUIFER TEST
PT - PACKER TEST

Figure 5-1 Example Piper Diagram of Cations and Anions In Recharge and Native Groundwater Samples from Nansemond WWTP Hampion Roads Sanitation District Virginia Beach, VA

Ch2m:

Figure 5-1. Example Piper Diagram of Cations and Anlons In Recharge and Native Groundwater Samples from Nansemond WWTP

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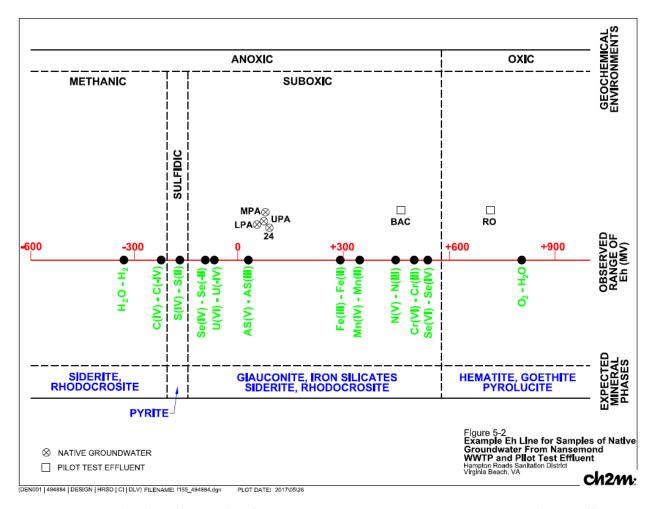


Figure 5-2. Example Eh Line for Samples of Native Groundwater From Nansemond WWTP and Pilot Test Effluent

### 5.4 Evaluate Attenuation Mechanisms

Identifying attenuation mechanisms for an individual constituent can represent the most challenging aspect of evaluating column or field scale data. HRSD should evaluate attenuation mechanisms for constituents exhibiting a partial to full breakthrough curve. The assessment of the breakthrough curve should include a working knowledge of the geochemical environment, displayed in the column or sand interval of the PAS. Evaluating attenuation mechanisms for individual constituents will also require reviewing available literature describing similar studies.

## 5.5 Solute Transport and/or Geochemical Modeling

Solute transport, reactive transport, and geochemical equilibrium models provide powerful tools for evaluating the migration of constituents in groundwater. They achieve their greatest level of effectiveness with accurate input data, and the analysts' understanding of the geochemical environment undergoing modeling. The models are most often used to mathematically simulate the transport of a constituent in a column or an aquifer. Important input parameters are modified to duplicate the column or field scale results (calibration). Once a model is calibrated and verified, simulations are run to predict the migration of a constituent under differing conditions, or to extend the time scale beyond the duration of a column or field scale experiment.

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Solute transport models like MT3D (Zhang, 1990), MOC (Konikow et al., 1978), SUTRA (Voss et al., 2004), etc. use mathematical coefficients to represent hydrodynamic (advection, dispersion, and diffusion), solute decay, and chemical reactions. Single numerical terms representing complex environmental reactions are used as input to a solute transport model. Solute transport models allow simulating system in one, two, and three dimensions. They represent the most popular tools for simulating constituent transport in a column.

By comparison, geochemical models like MINTEQA2 (Allison, et. Al., 1991), PHREEQC (Parkhurst, 1995), and the Geochemist's Workbench (Bethke, 1996) use water chemistry, mineral phases, temperature, and the partial pressures of gas as input to simulate mineral speciation and solubility, surface complexation, mass transfer/reaction path, and inverse reactions in a beaker. Solute transport is often simulated in one dimension as migration along a line from a source. PHREEQC allows simulating differing geochemical environments along a linear flowpath, and reversing the direction of flow.

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# Deploying the Testing Results

This section describes applying the results and conclusions from the column and field scale SAT testing to assess groundwater quality, analyze risk to local groundwater users, and how to apply the results to obtain regulatory benefits for an advanced wastewater treatment facility. At this stage of the project, these topics seem highly speculative and will require revision as testing proceeds at SWIFTRC.

## 6.1 Assessing the Influence of Managed Aquifer Recharge Operations on Groundwater Quality

At the end of column and field scale testing, HRSD will possess significant amounts of data that support determining how MAR operations will influence groundwater quality in the PAS. Where the monitoring described in Attachment C will focus on cations, anions, trace metals, and other general water quality parameters, this assessment will focus on constituents inherent with advanced treatment of wastewater including pathogens, TOC/DOC, nitrogen species, CECs and NDMA.

## 6.2 Analysis of Risk to Local Receptors

One application of the column and field scale testing should involve simulating the migration of a constituent(s) that exhibited breakthrough during testing, toward the closest receptors using groundwater from the PAS. Receptors could include large (municipal, industrial, agricultural, etc.) and small scale users (domestic) to assess a constituent migration under ambient and strong pumping gradients. This type of analysis should include conservative simulations, involving only advective transport (particle tracking), in combination with simulations comprising the geochemical and/or biological factors that attenuate constituents during migration. If testing demonstrates, that SAT removes or acceptably attenuates the constituents of interest, HRSD may not require these analyses.

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